Department of Chemical Engineering

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Incorporation of CIGS Cells into Photo-Electrochemical Reactors...

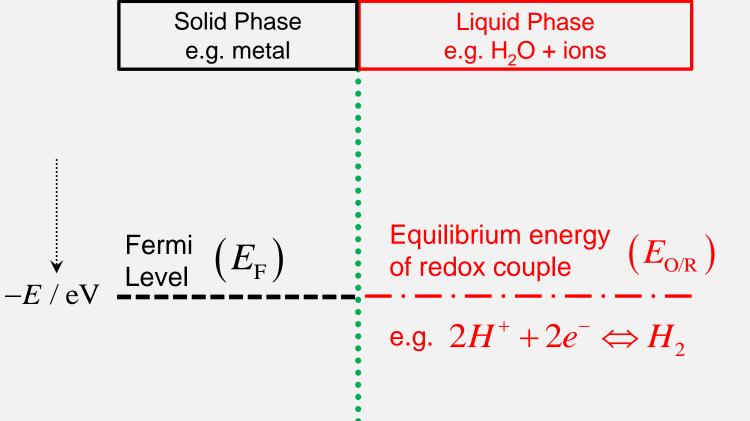
Anna Hankin

Postdoctoral Research Associate

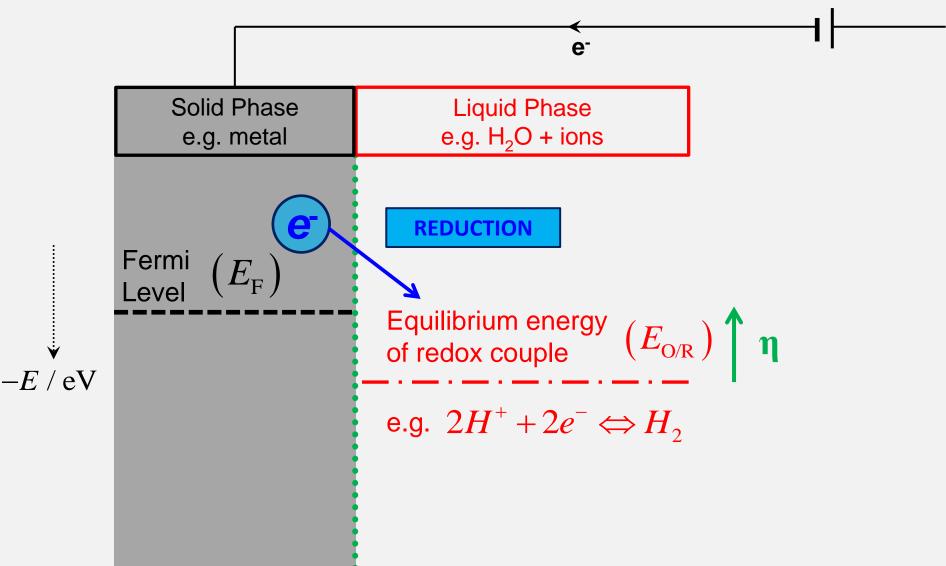
Electrochemical Engineering Research Group

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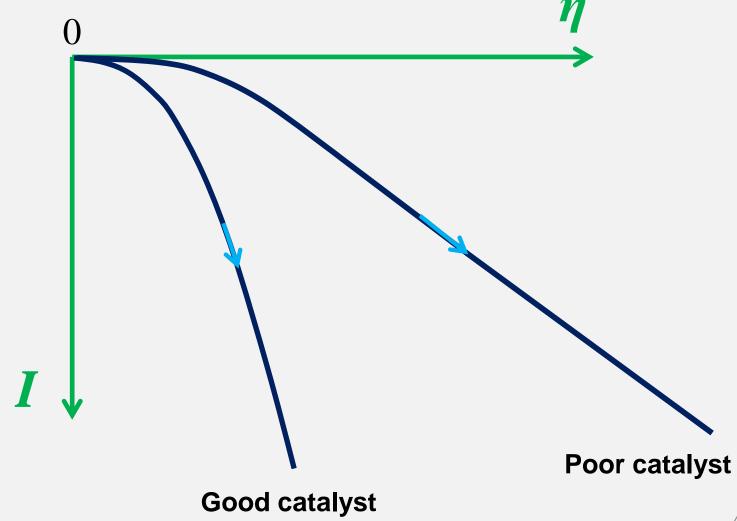
Principles of electrochemical reactions



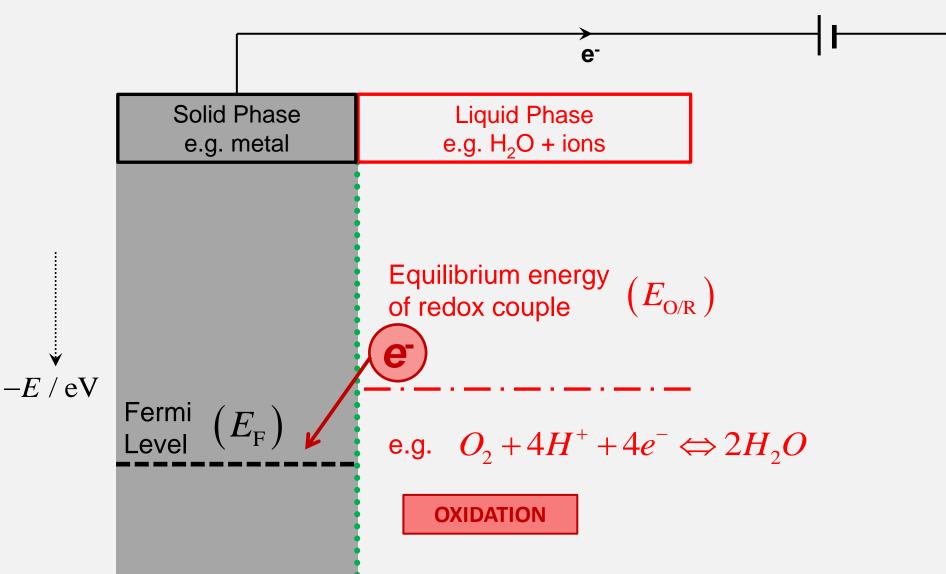
Principles of electrochemical reactions



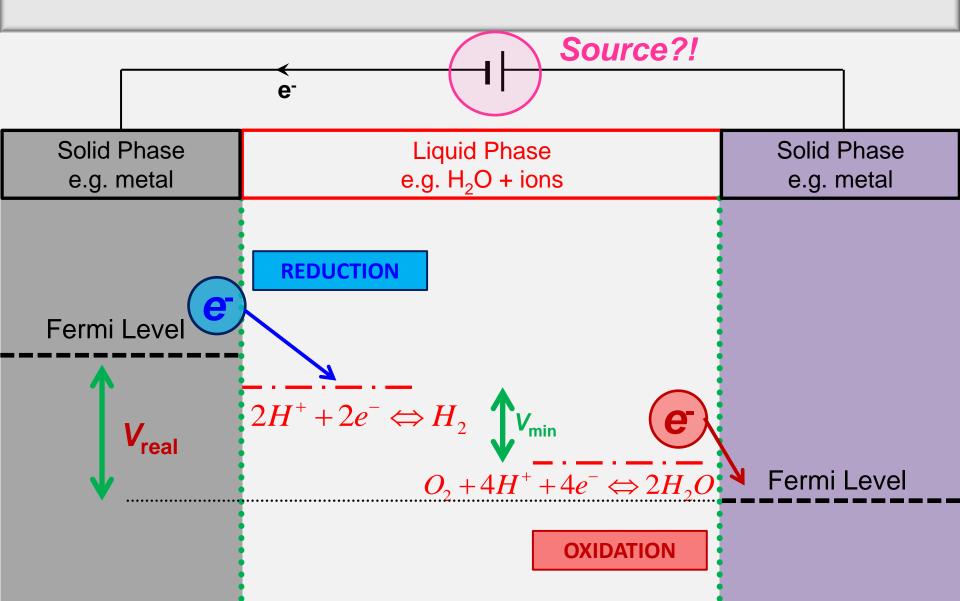
Kinetics



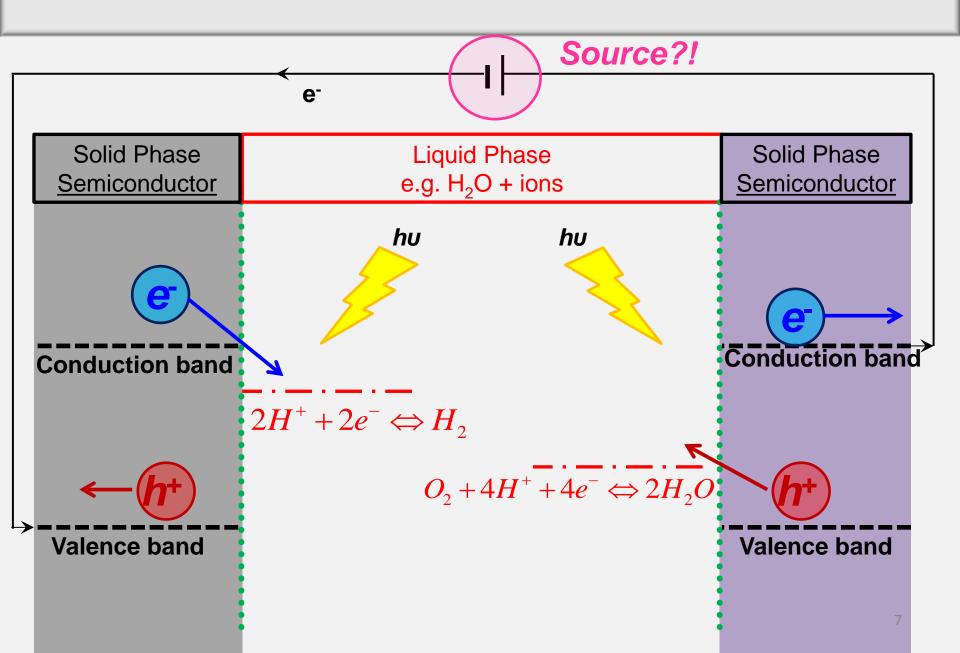
Principles of electrochemical reactions



Principles of electrochemical reactors



Electrochemical to Photo-Electrochemical



Electrode Design

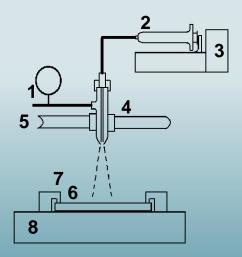
Principal requirements for a photo-electrode:

- Good photo-absorber
- Suitable conduction and valence band energies
- Suitable direction of internal bias
- Chemically robust
- Cheap

Hematite Photo-Anode Production

Spray Pyrolysis Setup

- 1. Compressed Air
- 2. Precursor reservoir
- 3. Syringe pump
- 4. Quartz nebuliser
- 5. CNC machine
- 6. Substrate
- 7. Clamping block
- 8. Hotplate

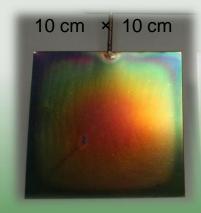


Fe₂O₃ coatings produced by nebulising Fe^{III}Cl₃ in solvent onto heated substrate

$Ti \mid Fe_2O_3-SnO_2$

Variables:

- Dopants → Sn^{IV} (0.6 %) (increase photocurrent)
- Substrate → Ti, FTO
 (flexibility with illumination)

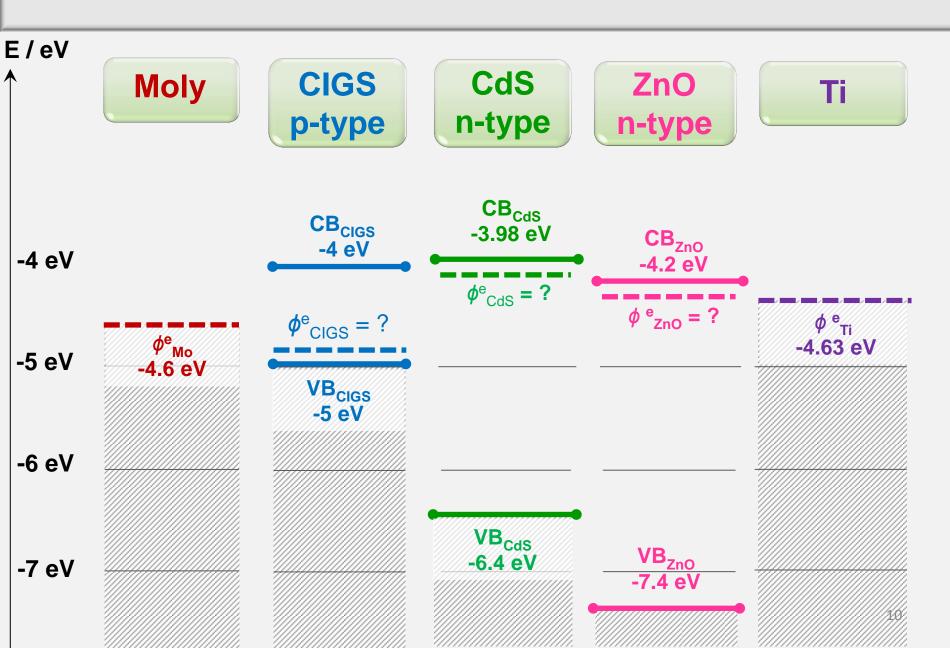


Photographic image

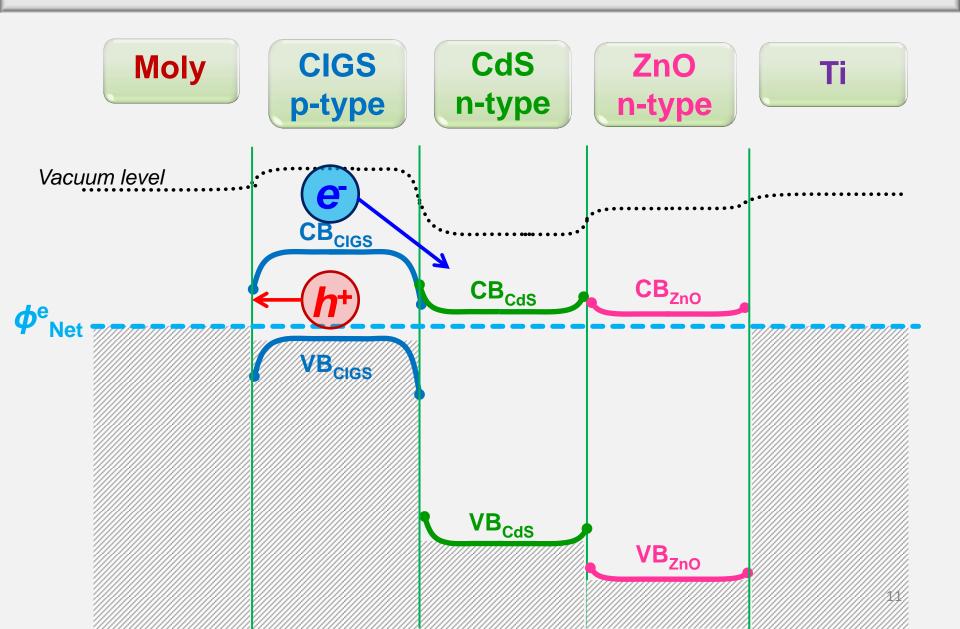


SEM image

The Mo | CIGS | CdS | ZnO | Ti system

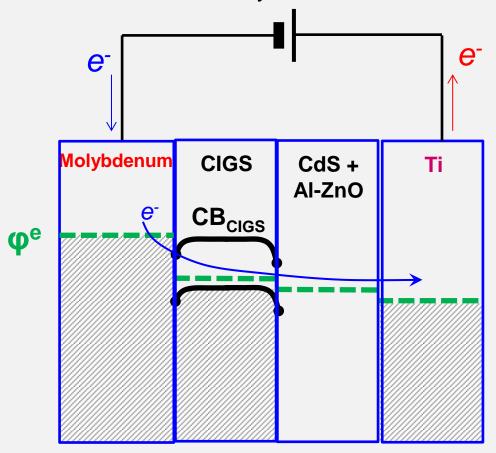


The Mo | CIGS | CdS | ZnO | Ti system



The Mo | CIGS | CdS | ZnO | Ti system

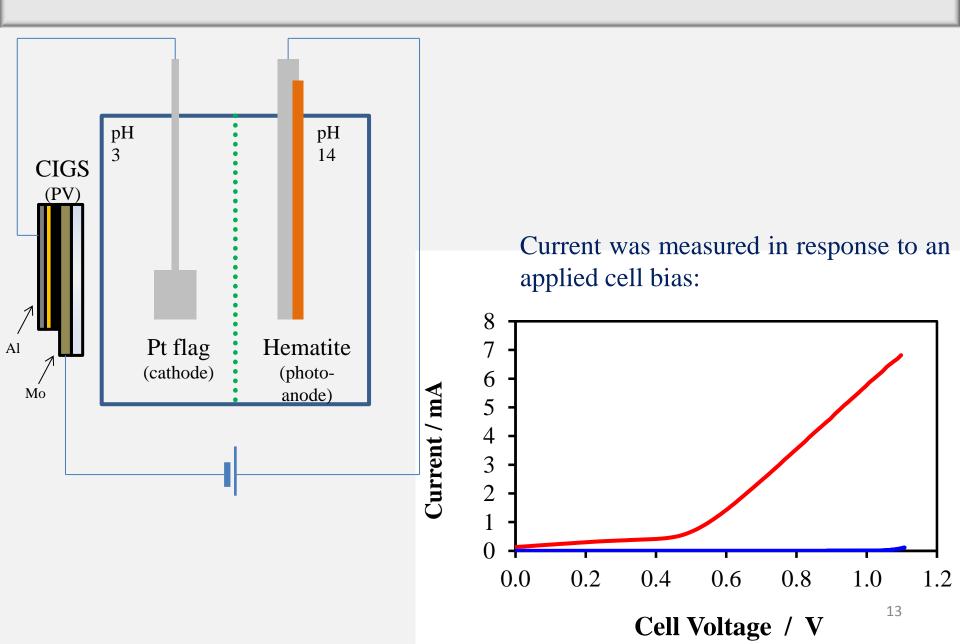
To replicate the effect of illumination on energy levels within the photo-cathode, the photo-cathode (when studied in the air) must be biased in this way:



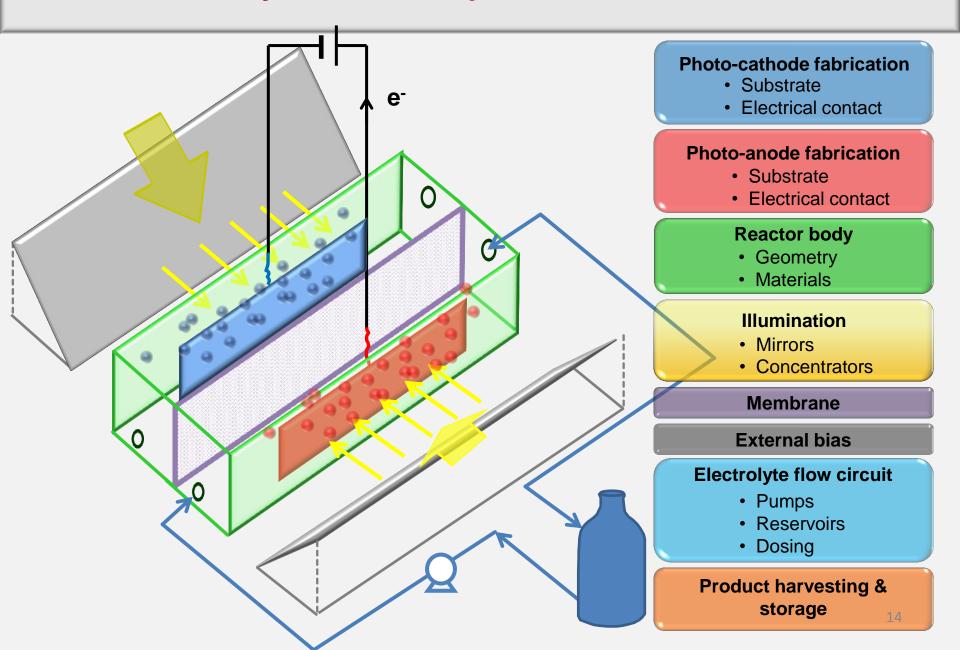
For the physicists:
This is 'reverse bias'?

For the electrochemists: Dark current and photocurrent refer to the flow of electrons in this direction.

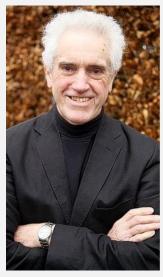
Performance of AI coated CIGS Cells



Device & System Development



The London Team



Keith Barnham (Physics)



Geoff Kelsall (Chem. Eng.)



Amanda Chatten (Physics)



Anna Hankin (Chem. Eng.)



José Videira (Physics)