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Synthesis of Azadirachtin *via* Relay Route

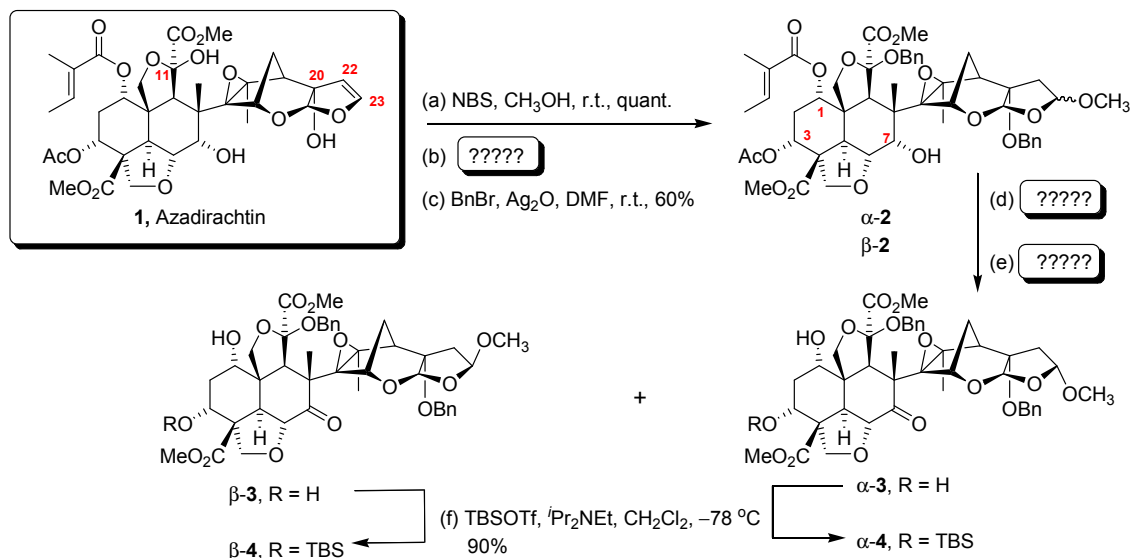
Indian neem tree, *Azadirachta indica* is renowned for its antiseptic, antiviral, antifungal, antibacterial, anti-infertility, antipyretic, anti-inflammatory, anti-ulcer, antihelmintic and insect repellent properties for several thousands of years and is hailed as “Village Pharmacy”, “Divine Tree”, “Heal All”, “Nature’s Drugstore” and “Panacea for all diseases” across many south Asian countries.¹ Crude neem tree extracts have been traditionally used as “eco-friendly” insect repellants, pesticide and fungicide in agriculture across the Indian subcontinent.² Azadirachtin **1** is a structurally complex natural product first derived from the Indian neem tree in 1968,³ has remained to be of interest among the organic chemists, biochemists and agro chemists owing to its potent antifeedant and growth-disruptant properties against a broad spectrum of insect species whilst displaying extremely low mammalian toxicity [LD₅₀(rat)>5 g/Kg] and appears to cause little disruption to beneficial species such as pollinating bees and ladybirds.⁴ After several years of intense research the structure of Azadirachtin was elucidated in 1987.⁵

For decades the total synthesis of azadirachtin **1** has lingered to be a tougher nut to crack, particularly because of the complexity involved in its structure arising from an array of sixteen contiguous stereogenic centres, seven of which are tetrasubstituted along with a diverse assortment of several oxygenated functionalities. Adding to challenges posed by the structural complexity, azadirachtin **1** was prone to rearrangement under acidic, basic or photolytic conditions. Also, presence of influential intramolecular hydrogen-bonding interactions, evident from X-ray crystallography studies,⁶ dictated the conformation and reactivity of the azadirachtin **1**.⁷

The synthesis of azadirachtin **1** was pursued simultaneously by several groups in the world including that of Ley,^{7,8} Nicolaou,⁹ Watanabe.¹⁰ It was only recently Ley’s group overwhelmed the synthesis of Azadirachtin **1** *via* a “Relay Route” protocol.^{7,8} This landmark synthesis featured a total of 64 meticulous steps, engaging over 40 synthetic chemists in the Ley group for over 22 years.¹¹ This problem session will feature analysis of this monumental synthesis from Ley’s group.

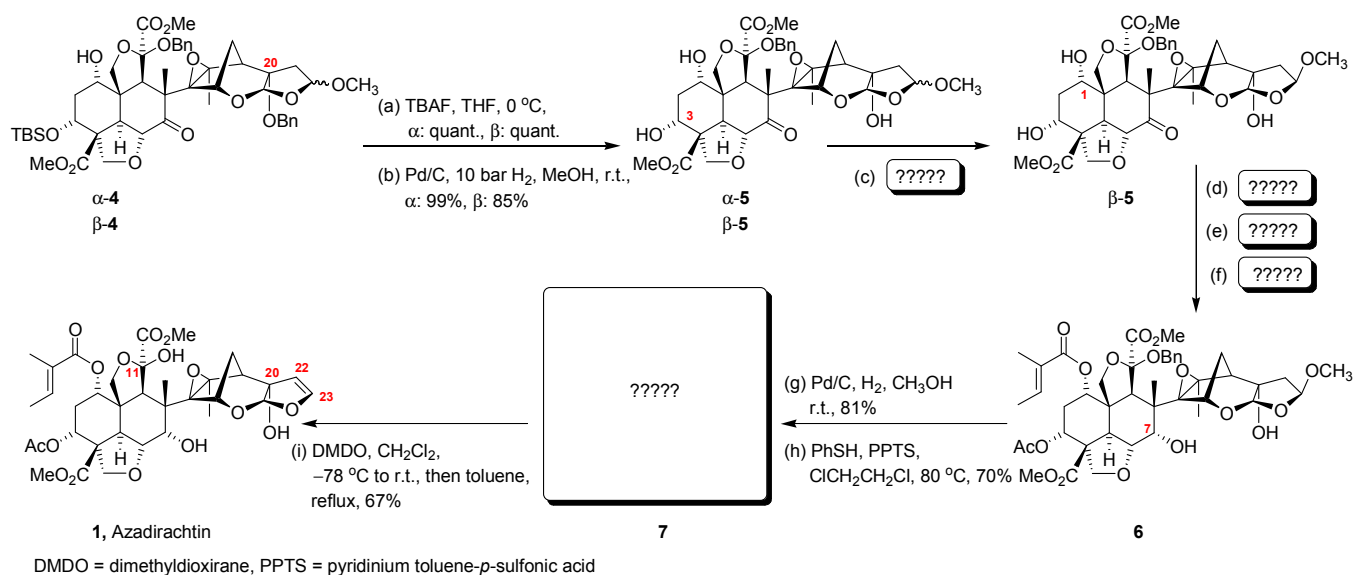
PART-1: A Relay Route for the Synthesis of Azadirachtin **1**

Scheme 1: Degradation of Azadirachtin **1**

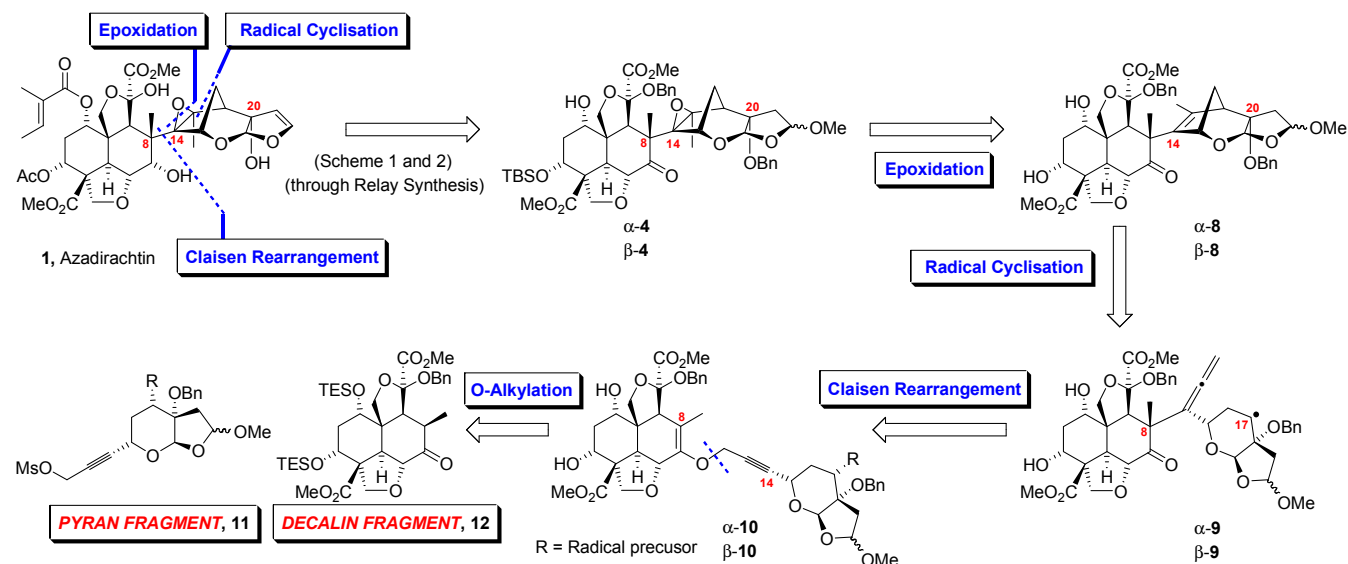


Questions for Discussion (for Scheme 1):

- (1) What is meant by "Relay Synthesis"? In what way does it differ from the conventional convergent synthesis and diversity-oriented synthesis (DOS)?
- (2) In the first step of the degradation of azadirachtin **1**, why is the alkene in the dihydrofuran moiety react preferentially with the NBS while the other alkene does not react at all?
- (3) Suggest a suitable reagent for step (b).
- (4) Discuss why there is selective protection of the hydroxyl groups at C11 and C20 to afford dibenzyl ether **2** [in step (c)]?
- (5) Suggest suitable reagents and conditions for steps (d) and (e).

Scheme 2: Return of α -4 and β -4 to Azadirachtin 1**Questions for Discussion (for Scheme 2):**

- (6) How will you account for the selective debenzylation of OBn group to free alcohol using H₂ gas on Pd/C in methanol [from **4** to **5** in step (b)]. (The authors report that this step was extensively optimised by screening various solvents and varying other experimental conditions without much explanation).
- (7) Since the authors successfully achieved the conversion of C23 β epimer of intermediate **5** to azadirachtin **1** in a first generation relay approach,¹² they intended to epimerize the C23 α epimer of the intermediate **5** to the corresponding β epimer. What reagents/conditions could have they used to achieved this? [step (c)]. In this step, the C23 α epimer of the intermediate **5** was successfully epimerised into a 1:1 mixture of diastereoisomers, from which the desired epimer (β -5) was isolated cleanly.
- (8) Propose suitable reagents and conditions to effect conversion of β -5 to **6** [three steps - step (d), (e) and (f)]. What will be your order of events?
- (9) What is the structure of the intermediate **7**?
- (10) How is dimethyldioxirane prepared? Give mechanism for the transformation of **7** to the desired natural product azadirachtin **1** [step (i)].

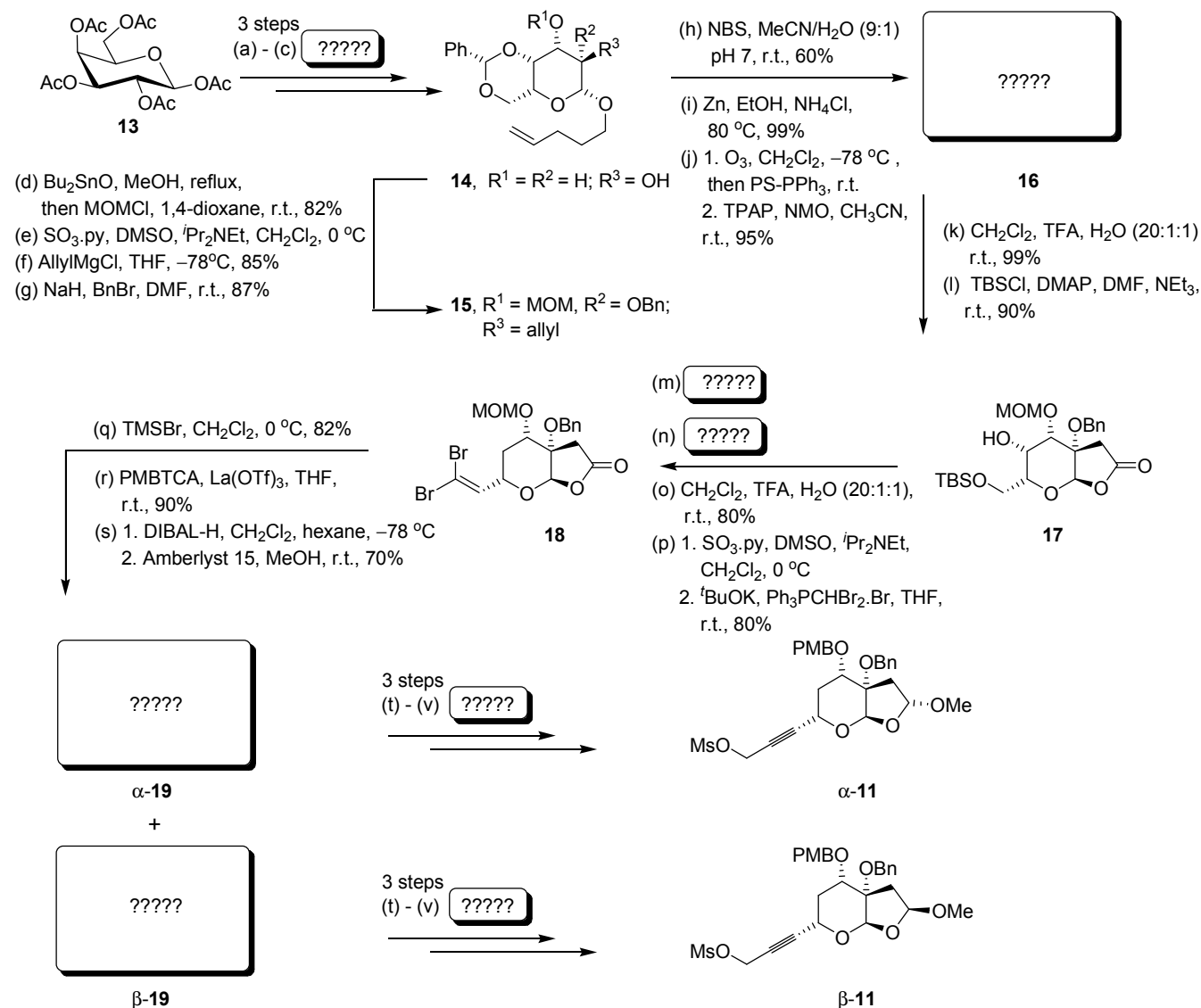
PART-II: Synthesis of the EPIMER β -4**Scheme 3: Retrosynthetic Analysis**

Bn = Benzyl; Ms = methanesulfonyl; TBS = *tert*-butyldimethylsilyl; TES = triethylsilyl

Questions for Discussion (for Scheme 4):

- (11) The authors synthesized the diol **14** starting from β -D-galactose pentaacetate **13** in three simple steps. Suggest them. Can you also work out an alternative synthesis of the diol **14** starting from simple and commercially available synthons and catalysts.
- (12) Taking into consideration of the reagents used in steps (h) – (j) predict the structure of the compound **16**. Also, rationalise the reaction mechanism involved in step (h) where NBS is used as the reagent.
- (13) What is TPAP? How does NMO act as a co-oxidant during TPAP oxidation? What other co-oxidants will you use instead of NMO?
- (14) What are the reactions, reagents and conditions involved in steps (m) and (n)? What are the names of the reactions involved in these steps?
- (15) How will you alkylnate the epimers **19** to afford the propargylic mesylate **11** [in three steps (t) – (v)]. [At this stage, the authors had no firm decisions as to whether one or both of these distereoisomers would be brought forward to the natural product. In the end both served as viable precursors (see next section later)].
- (16) Imagine if you were involved in the synthesis of the pyran **11**, how will you distinguish between the two epimers of **11** by means of physical methods, say by NMR spectroscopic analysis?

Scheme 4: Synthesis of Pyran Fragment 11



MOMCl = Chloromethyl methyl ether; py = pyridine; DMSO = dimethyl sulfoxide; Bn = Benzyl; NBS = *N*-bromosuccinimide; NMO = *N*-methylmorpholine-*N*-oxide; PS = polymer support; DMAP = 4-(*N,N*-dimethylamino)pyridine; DMF = *N,N*-dimethyl formamide; TBSCl = *tert*-butyldimethylsilyl chloride; TMS = trimethylsilyl; PMBTCA = *p*-methoxybenzyl trichloroacetimidate; DIBAL = *di*sobutylaluminium hydride; TFA = trifluoroacetic acid.

Questions for Discussion (for Scheme 5):

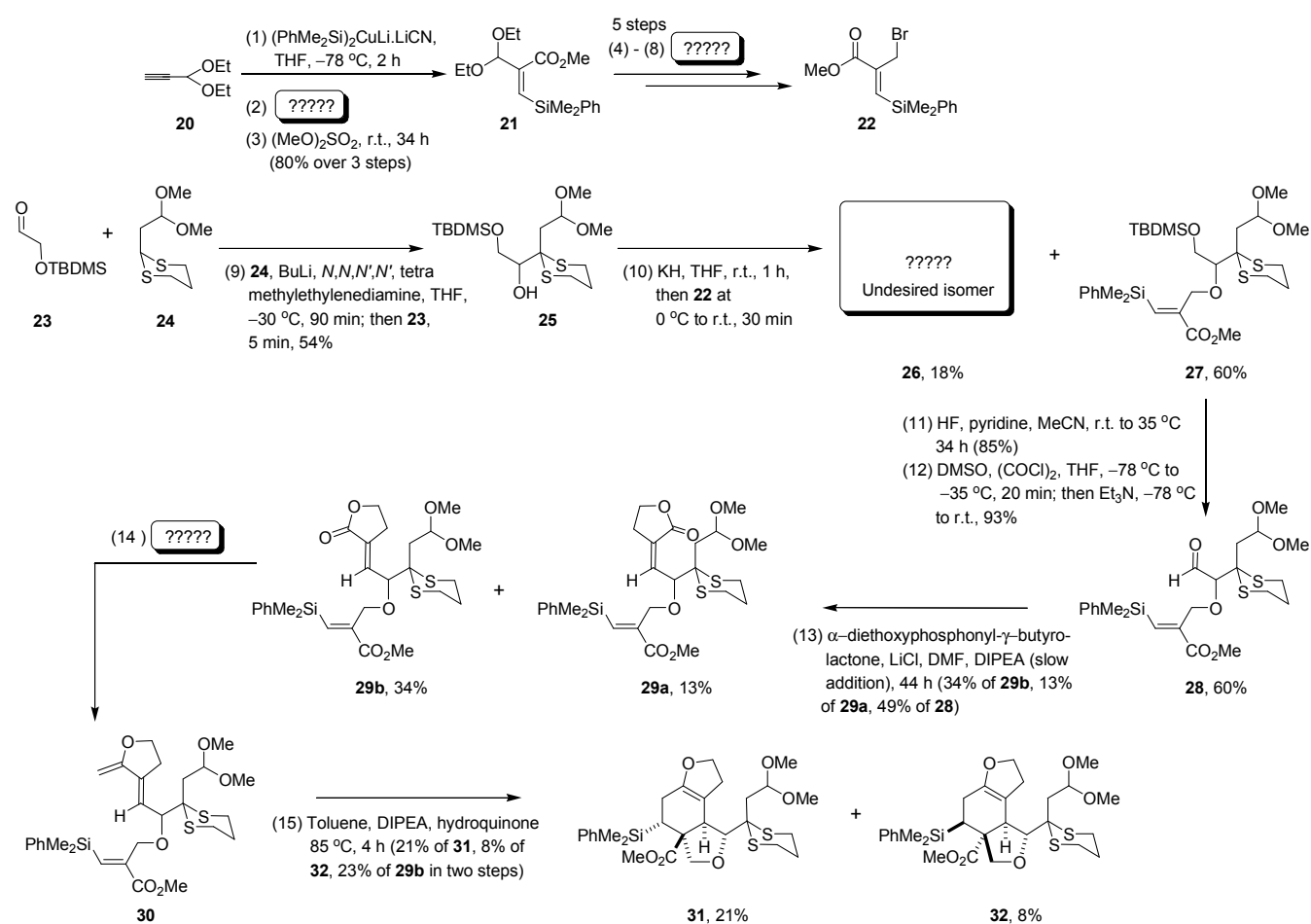
Ley and co-workers reported the multi-step synthesis of the decalin fragment **12** in 1992.¹³

- (17) Why is the addition of Fleming's dimethyl(phenyl)silylcuprate on to the propargylic acetal **20** stereospecific and regioselective [step (1), Scheme 5]?
- (18) What carboxylation reagents and conditions will you suggest for step 2 (Scheme 5) which proceeds soon after the successful addition of the dimethyl(phenyl)silylcuprate on to the propargylic acetal **20**?
- (19) Propose suitable reagents and conditions for steps (4)-(8) (Scheme 5) in order to obtain the bromo ester **22** from enoate **21**.

Questions for Discussion (for Scheme 5) contd.:

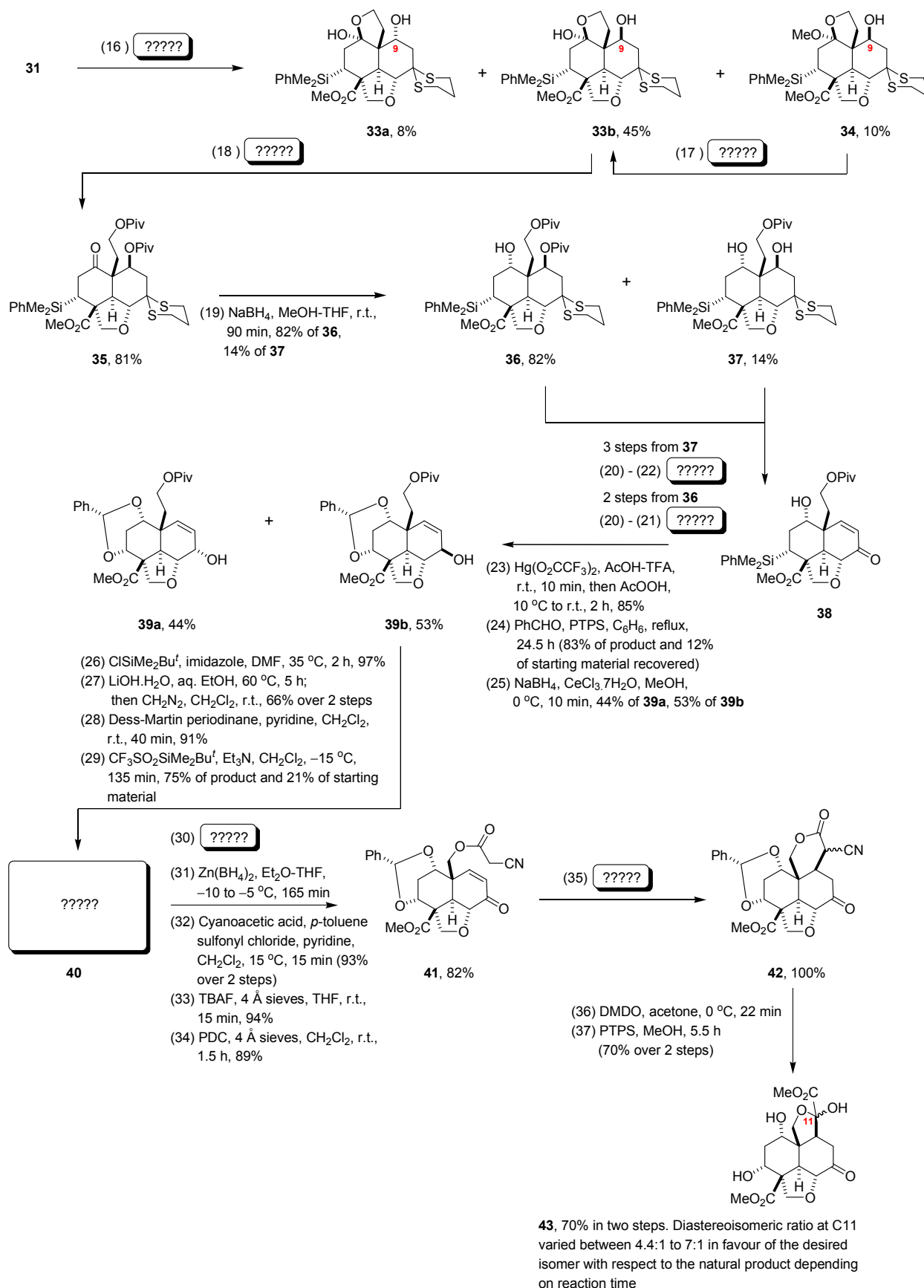
- (20) What is the role of TMEDA in step (9) involving addition of the anion derived from the dithiane **24** to TBDMS-protected hydroxy aldehyde **23**?
- (21) While coupling of the alcohol **25** with the bromo ester **22** was attempted using KH as the base [step (10), Scheme 5], the reaction yielded the desired product **27** in 60% yield along with an undesired product **26** (an isomer of **27**) in 18% yield. What could be the structure of the undesired compound **26** and outline how could have it be formed?
- (22) What is the name of the olefination reaction involved in step (13) and explain the reason for the selectivity observed for the formation of the alkene **29b** as the major product from this reaction?
- (23) Speculate suitable reagents and conditions for step (14). [The authors did not purify and isolate the product **30** and carried on to the next step (15) directly]
- (24) Which of the compounds **31** and **32** are *exo* and *endo* product(s) [step (15), Scheme 5]? What are the roles of hydroquinone and DIPEA in this reaction? Rationalise why the intramolecular Diels-Alder reaction of **30** has yielded product **31** as the major product? How can the isomers **31** and **32** be distinguished by NMR spectroscopic analysis?

Scheme 5: Synthesis of Decalin Fragment 12 (part-I):



DMF = *N,N*-Dimethyl formamide; DIPEA = *N,N*-diisopropylethylamine

Scheme 6: Synthesis of Decalin Fragment 12 (part II) Contd.:

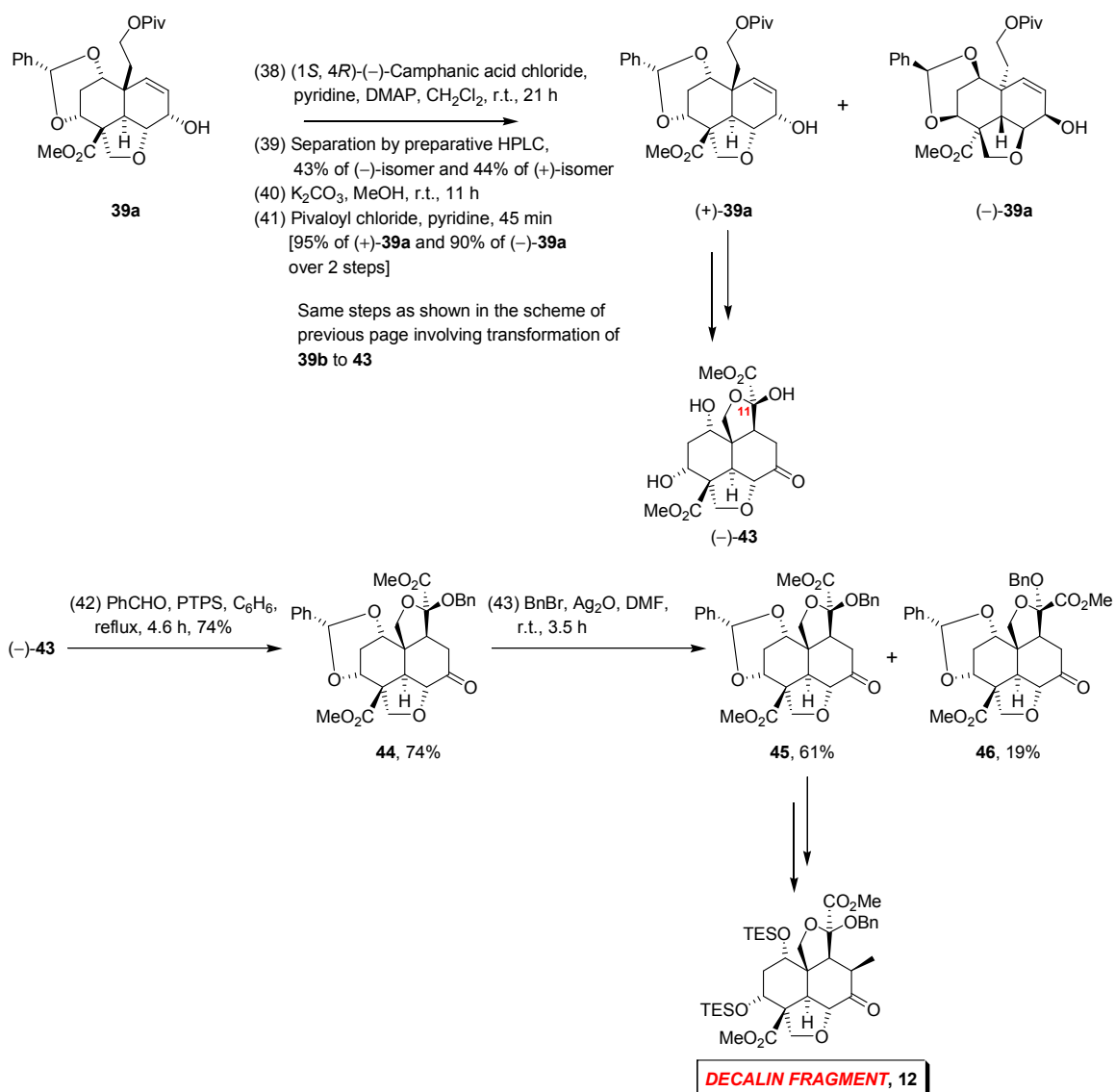


TFA = Trifluoroacetic acid; PTPS = pyridinium tosate; DMF = *N,N*-Dimethyl formamide; TBAF = tetra-*N*-butylammonium fluoride; Å = ångstrom; PDC = pyridinium dichromate; DMDO = dimethyldioxirane

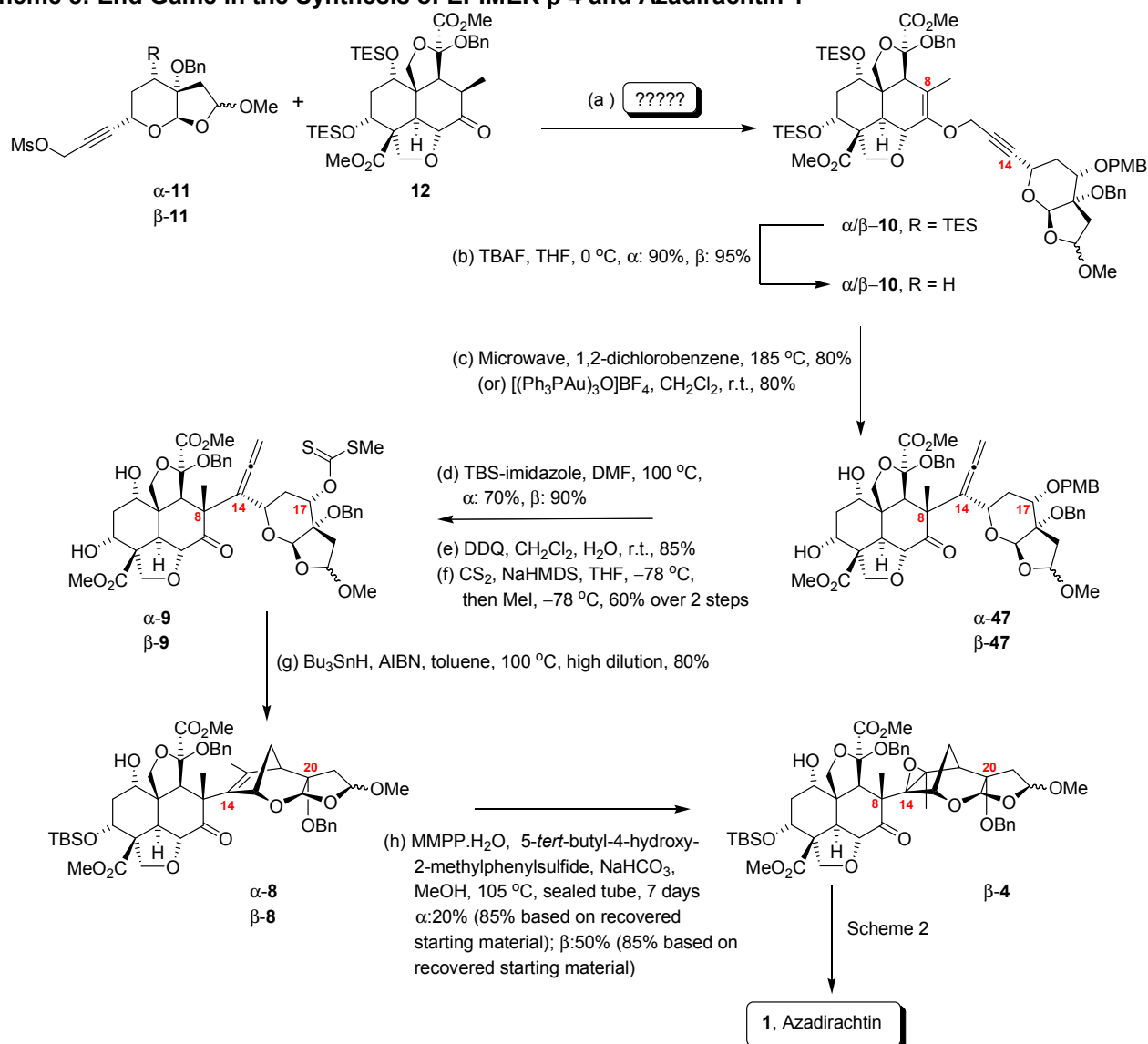
Questions for Discussion (for Scheme 6 and 7):

- (25) What reagent and conditions can be used for the intramolecular cyclisation of the compound **31** to afford the products **33a**, **33b** and **34**? What is the name of the reaction that is involved in this step [step (16), Scheme 6]?
- (26) How is the compound **34** converted to the compound **33b** [step (17), Scheme 6]?
- (27) Propose two and three step route to **38** from **36** and **37** respectively [steps (20)-(22), Scheme 6].
- (28) Explain the protocol involved in step (23) [Scheme 6] where phenyltrialkylsilane **38** is treated with mercury(II) trifluoroacetate with acetic acid and trifluoroacetic acid mixture?
- (29) Based on the steps (26) to (29) [Scheme 6], predict the structure of the compound **40**.
- (30) Propose suitable protocols for step (30) and step (35) [Scheme 6].
- (31) Explain the mechanisms involved in steps (36) and (37) [scheme 6] in order to afford the decalin fragment **43** from the cyano lactone **42**.
- (32) What is the structure of (1*S*,4*R*)-Camphanic acid? Name the protocol associated with the step (38) [Scheme 7].

Scheme 7: Synthesis of Decalin Fragment 12 (part III) Contd.:



DMAP = 4-(*N,N*-Dimethylamino)pyridine; HPLC = high performance liquid chromatography; PTPS = pyridinium tosylate; DMF = *N,N*-Dimethylformamide; Bn = benzyl

Scheme 8: End Game in the Synthesis of EPIMER β -4 and Azadirachtin 1

DDQ = Dichlorodicyanoquinone; TBAF = tetra-*N*-butylammonium fluoride; HMDS = hexamethyldisilazane; AIBN = azobisisobutyronitrile; TES = triethylsilyl; TBS = *tert*-butyldimethylsilyl; THF = tetrahydrofuran

Questions for Discussion (for Scheme 8):

- (33) How will you couple the the pyran fragment **11** with the decalin fragment **12** [step (a), Scheme 8]? The authors used ten fold excess of decalin **12** for complete conversion to propargylic enol ether **10**.
- (34) What type of rearrangement is involved in the conversion of the propargylic enol ether **10** to the allene **47** [Scheme 8]? This rearrangement, which constructs the key C8-C14 bond whilst simultaneously installing the requisite allene, can be induced thermally or by gold(I) catalysed conditions – discuss in detail on the mechanisms involved in this reaction under both conditions.
- (35) Give a reasonable mechanism involved in step (g) [Scheme 8] involving tributyltin hydride and AIBN.
- (36) What is MMPP [step (h), [Scheme 8]]? Speculate the role of 5-*tert*-butyl-4-hydroxy-2-methylphenylsulfide in this reaction. The authors quote that this epoxidation reaction of both α - and β -**8** yielded the β diastereoisomer of the relay target (β -**4**), implying an epimerisation of alkene α -**8** to its β -**8** form prior to epoxidation. This hypothesis was further supported by the observation of the lower yield from the transformation of α -**8**.

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